## A STEREOCONTROLLED TOTAL SYNTHESIS OF OPTICALLY ACTIVE (R,R)-PHYTOL

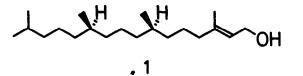
Tamotsu Fujisawa,\* Toshio Sato, Tatsuo Kawara, and Kazuo Ohashi Chemistry Department of Resources, Mie University, Tsu, Mie 514, Japan

Summary: A stereospecific synthesis of (R,R)-phytol with highly stereochemical purity in both absolute and geometrical configurations was achieved by utilizing the  $S_N2$  type ring-opening reaction of (R)- $\beta$ -methyl- $\beta$ -propiolactone, and the  $S_N 2^{\prime}$  type ring-opening reaction of isopropenyl oxirane, from (R)-pulegone as a starting material.

Phytol (1) occurs abundantly in plants as an alcohol part of the ester sidechain of chlorophyl, and is a useful material for the syntheses of  $\alpha$ -tocopherol (vitamin E)<sup>1</sup> and phylloquinone (vitamin  $K_1$ ).<sup>2</sup> Natural phytol is an acyclic diterpene alcohol possesing an R-configuration at two asymmetric carbons (C7 and  $C_{11}$ ) and an E-configuration of the double bond.<sup>3</sup>

In spite of the enormourous literatures on the synthesis of racemic phytol,<sup>4</sup> a few reports have been published on that of optically active phytol.<sup>3,5</sup> We wish to describe here a method for the stereocontrolled synthesis of (R,R)-phytol (1) with highly optical purity via (3R,7R)-3,7,11-trimethyldodecanoic acid (2); one asymmetric center at the  $C_{11}$  carbon of 1 was derived from easily available (R)-(+) pulegone (4) and the other one at the  $C_7$  carbon was constructed by the  $S_N 2$  type ring-opening reaction of  $(R)-(+)-\beta$ -methyl- $\beta$ -propiolactone (10) with a Grignard reagent in the presence of a copper(I) salt, which proceeds with almost complete inversion.<sup>6</sup> In addition, the geometrical configuration at the prenyl alcohol moiety was controlled by the  $S_N2'$  type reaction of isopropenul oxirane (15) with a Grignard reagent in the presence of a copper(I) salt.<sup>7</sup>

The acyclic isoprenoid synthons of C15 unit possessing two chiral centers, such as 2 and 3, are important intermediates for the synthesis of 1. Recently, in connection with the synthesis of  $\alpha$ -tocopherol many efforts have been devoted



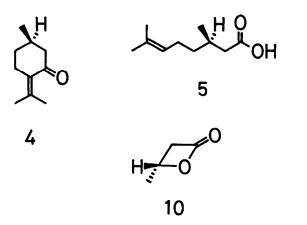
**2**  $X = CO_2H$  **3**  $X = CH_2OH$ 

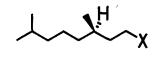
to the synthesis of 2 or 3 from several chiral synthons, prepared by microbiological procedures,<sup>8</sup> classical optical resolution of racemic mixtures,<sup>9</sup> or from naturally occurring chiral compounds.<sup>10</sup> In the present synthesis, one of the key intermediates of  $C_{11}$  unit, (R)-4,8-dimethylnonyl bromide (9), easily prepared from 4, was converted to the Grignard reagent (11) which was coupled with the lactone 10 of C<sub>4</sub> unit to afford the C<sub>15</sub> acid 2 (*vide infra*).

According to the procedure by Overberger et al.,<sup>10</sup> 4 was converted into optically pure (R)-(+)-citronellic acid (5) in 98% yield;  $[\alpha]_{D}^{23} + 9.08^{\circ}$  (C 3.17, MeOH), lit.<sup>11</sup>  $[\alpha]_{D} + 9.05^{\circ}$  (C 3.2, MeOH). Platinum-catalyzed hydrogenation of 5 and subsequent reduction with lithium aluminum hydride gave (R)-(+)-3,7-dimethyloctanol (6) in 99% yield. Treatment of 6 with hydrogen bromide at 115 °C afforded (R)-(-)-3,7-dimethyloctyl bromide (7) in 99% yield; bp 68~70 °C/1.5 mmHg;  $[\alpha]_{D}^{23} - 6.56^{\circ}$  (neat). Cyanation of  $7^{12}$  and subsequent alkaline hydrolysis furnished (R)-(-)-4,8-dimethylnonanoic acid (8) in 82% yield; bp 120 °C/2 mmHg;  $[\alpha]_{D}^{23} - 0.59^{\circ}$  (neat). Reduction of 8 with lithium aluminum hydride followed by bromination with hydrogen bromide gave (R)-(-)-4,8-dimethylnonanoyl bromide (9) in 92% yield; bp 75~80 °C/1 mmHg;  $[\alpha]_{D}^{23} - 2.55^{\circ}$  (neat). Thus, the chiral synthon of  $C_{11}$  unit, which possesses one asymmetric carbon corresponding to the C<sub>11</sub> carbon of 1, was obtained in an overall yield of 72% from 4.

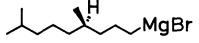
The chiral C<sub>4</sub> unit (R)-lactone 10, which involves the C<sub>7</sub> asymmetric carbon of 1, was easily obtained from optically pure (S)-(+)-3-bromobutyric acid ([M] $_{546}^{24}$ +116.2° (0.168 M, 2M HClO<sub>4</sub>), lit.<sup>13</sup> [M] $_{546}^{25}$ +116.5° (0.163 M, 2M HClO<sub>4</sub>)), in a yield of 71%; [ $\alpha$ ] $_D^{24}$ +26.8° (C 5.0, CHCl<sub>3</sub>).<sup>6,14</sup>

Synthesis of the key intermediate 2 was easily achieved by the  $S_N 2$  type ringopening reaction of the lactone 10 with the Grignard reagent 11, prepared from 9 in 95% yield, in the presence of a copper(I) salt. Thus, when 11 was added to a mixture of 10 and copper(I) iodide (2 mol%) in THF-Me<sub>2</sub>S (20:1) at -20 °C and the reaction mixture was allowed to warm to room temperature for 2.5 h, 2 was





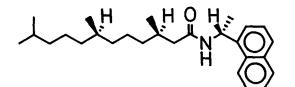
6 X = OH 8 X =  $CO_2H$ 7 X = Br 9 X =  $CH_2Br$ 



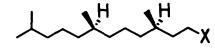
obtained in 90% yield; bp 180 °C (bath temp.)/1 mmHg;  $[\alpha]_D^{24}$  +5.46° (C 5.0, CHCl<sub>3</sub>), lit.<sup>15</sup>  $[\alpha]_D$  +5.43° (C 5.0, CHCl<sub>3</sub>). The specific rotation indicated high optical purity of 2, and the enantiomeric purity of the asymmetric C<sub>3</sub> carbon was determined by TLC analysis of the diastereomeric amide (12), prepared from the acyl chloride of 2 and (R)-(+)- $\alpha$ -(l-naphthyl)ethylamine. Separation of 12 by silica gel TLC (C<sub>6</sub>H<sub>6</sub>:AcOEt = 8:1) gave two components in a ratio of 96:4, corresponding to the (3R,7R)- and (3S,7R)-amides,respectively. Thus, the configuration of the  $\beta$ -carbon of (S)-3-bromobutyric acid was transformed into 2 with net retention of 96% through the two step inversion (cyclization to 10 and the ringopening reaction of 10). Since the transformation of 4 to 2 seemed to proceed without racemization, the enantiomeric purity of the asymmetric C<sub>3</sub> and C<sub>7</sub> carbon of 2 could be determined to be 96% and 100% R, respectively.

Stereoselective introduction of a prenyl alcohol moiety with an E-configuration was readily achieved by a copper-catalyzed  $S_N2'$  type reaction of Grignard reagent with the oxirane  $15.^{7,16}$  The acid 2 upon reduction with lithium aluminum hydride gave the alcohol 3, followed by the treatment with hydrogen bromide to afford the corresponding bromide 13 in 87% yield; bp 132 °C/0.5 mmHg;  $[\alpha]_D^{24}$  -4.44° (C 5.0, CHCl<sub>3</sub>). The Grignard reagent (14), prepared from 13 and magnesium by refluxing in THF in 76% yield, was added to a solution of 15 and copper(I) iodide (5 mol%) in THF-Me<sub>2</sub>S (20:1) at -20 °C. After the reaction mixture was allowed to warm to 0 °C for 2 h, phytol (1) was obtained in 95% yield in a ratio of E to Z as 97:3.<sup>17</sup>

As mentioned above, the synthesis of (R,R)-phytol with highly stereochemical purity with regard to both absolute and geometrical configurations could be achieved by utilizing both copper(I) catalyzed Grignard reactions; the  $S_N2$  type ring-opening reaction of (R)- $\beta$ -methyl- $\beta$ -propiolactone and the stereoselective  $S_N2$ ' type one of isopropenyl oxirane.



(3R,7R)-12



13 X = Br 14 X = MgBr

(3S, 7R) - 12



15

Acknowledgment: We wish to thank Mr. A. Nishizawa for his assistance in the experimental work. The present work was partially supported by a Grant-in-Aid for Scientific Research from the Ministry of Education of Japan.

## References and Notes

- 1. J. W. Scott, F. T. Bizzaro, D. R. Parrish, and G. Saucy, Helv. Chim. Acta, 59, 290 (1976); H. Mayer, P. Schudell, R. Rüegg, and O. Isler, *ibid.*, <u>67</u>, 650 (1963), and references cited therein.
- Y. Naruta, J. Org. Chem., <u>45</u>, 4097 (1980), and references cited therein.
   J. W. K. Burrell, R. F. Garwood, L. M. Jackman, E. Oskay, and B. C. L. Weedon J. Chem. Soc. (C), 1966, 2144.
- 4. K. Sato, S. Mizuno, and M. Hırayama, J. Org. Chem., <u>32</u>, 177 (1967); K. Sato and Y. Kurihara, Yuki Gosei Kagaku Kyokai Shi, 20, 824 (1962), and references cited therein.
- 5. P. Karrer, A. Geiger, H. Rentschler, E. Zbinden, and A. Kugler, Helv. Chim. Acta, <u>26</u>, 1741 (1943).
- 6. T. Sato, T. Kawara, A. Nishizawa, and T. Fujisawa, Tetrahedron Lett., 21, 3377 (1980).
- 7. Stereoselective reaction of diorganocuprate with 15 was reported; R. J. Anderson, J. Am. Chem. Soc., <u>92</u>, 4978 (1970). 8. C. Fuganti and P. Graselli, J. Chem. Soc., Chem. Commun., <u>1979</u>, 995; N. Cohen,
- W. F. Eichel, R. J. Lopresti, C. Neukom, and G. Sausy, J. Org. Chem., <u>41</u>, 3505
- (1976); R. Zell, Helv. Chim. Acta, <u>62</u>, 474 (1979). 9. K. Chan, N. Cohen, J. P. De Noble, A. C. Specian, Jr., and G. Saucy, J. Org.
- Chem., <u>41</u>, 3497 (1976). 10. B. M. Trost and T. P. Klun, J. Am. Chem. Soc., <u>103</u>, 1864 (1981); J. Takahashi, K. Morı, and M. Matsui, Agric. Biol. Chem., <u>43</u>, 1605 (1979). 11. C. G. Overberger and J. K. Weise, J. Am. Chem. Soc., <u>90</u>, 3525 (1968).

- 12. J. R. Ruhoff, Org. Synth., Coll. Vol. II. 292 (1950). 13. A. R. Olson and R. J. Miller, J. Am. Chem. Soc., <u>60</u>, 2687 (1938).
- 14. J. R. Shelton, D. E. Agostinı, and J. B. Lando, J. Polym. Sci., A-1, 9, 2789 (1971).
- 15. D. Valentine, Jr., K. K. Chan, C. G. Scott, K. K. Johnson, K. Toth, and G.
- Saucy, J. Org. Chem., <u>41</u>, 62 (1976).
  16. Although the oxirane 15 is a useful reagent, there are a few reports on the preparative method.<sup>7,18</sup> It was prepared by the modified procedure of Shanklin et al.<sup>19</sup> 3-Methyl-1-phenylthio-3-buten-2-ol, prepared from methacrylaldehyde and phenylthiomethyllithium, was treated with  $Et_3OBF_4$  in  $CH_2Cl_2$  at room

+ PhSCH<sub>2</sub>Li  $\longrightarrow$  SPh  $\frac{1) \text{ Et}_{3}\text{OBF}_{4}}{2) \text{ NaH}}$ 15

temperature for 12 h. The solvent was exchanged by diglyme, and then a solution of the sulfonium salt was added to a suspension of NaH in diglyme at 0 °C under vacuum (0.51 mmHg). The produced oxirane was collected in a trap cooled by liquid N<sub>2</sub>, and distillation afforded pure 15 in 29% yield; bp 84 °C. 17. The ratio of E:Z was determined by capillary glpc analysis (F.F.A.P. 50 m) of

- phytyl acetate, prepared from 1 and acetic anhydride in pyridine.
- 18. P. M. Savu and J. A. Katzenellenbogen, J. Org. Chem., <u>46</u>, 239 (1981).
- 19. J. R. Shanklin, C. R. Johnson, J. Allinger, and R. M. Coates, J. Am. Chem. Soc., <u>95</u>, 3429 (1973).

(Received in Japan 24 August 1981)