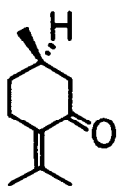


to the synthesis of 2 or 3 from several chiral synthons, prepared by microbiological procedures,⁸ classical optical resolution of racemic mixtures,⁹ or from naturally occurring chiral compounds.¹⁰ In the present synthesis, one of the key intermediates of C₁₁ unit, (R)-4,8-dimethylnonyl bromide (9), easily prepared from 4, was converted to the Grignard reagent (11) which was coupled with the lactone 10 of C₄ unit to afford the C₁₅ acid 2 (*vide infra*).

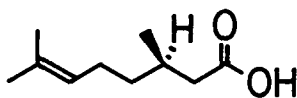
According to the procedure by Overberger et al.,¹⁰ 4 was converted into optically pure (R)-(+)-citronellic acid (5) in 98% yield; $[\alpha]_D^{23} +9.08^\circ$ (C 3.17, MeOH), lit.¹¹ $[\alpha]_D +9.05^\circ$ (C 3.2, MeOH). Platinum-catalyzed hydrogenation of 5 and subsequent reduction with lithium aluminum hydride gave (R)-(+)-3,7-dimethyloctanol (6) in 99% yield. Treatment of 6 with hydrogen bromide at 115 °C afforded (R)-(-)-3,7-dimethyloctyl bromide (7) in 99% yield; bp 68~70 °C/1.5 mmHg; $[\alpha]_D^{23} -6.56^\circ$ (neat). Cyanation of 7¹² and subsequent alkaline hydrolysis furnished (R)-(-)-4,8-dimethylnonanoic acid (8) in 82% yield; bp 120 °C/2 mmHg; $[\alpha]_D^{23} -0.59^\circ$ (neat). Reduction of 8 with lithium aluminum hydride followed by bromination with hydrogen bromide gave (R)-(-)-4,8-dimethylnonanoyl bromide (9) in 92% yield; bp 75~80 °C/1 mmHg; $[\alpha]_D^{23} -2.55^\circ$ (neat). Thus, the chiral synthon of C₁₁ unit, which possesses one asymmetric carbon corresponding to the C₁₁ carbon of 1, was obtained in an overall yield of 72% from 4.

The chiral C₄ unit (R)-lactone 10, which involves the C₇ asymmetric carbon of 1, was easily obtained from optically pure (S)-(+)-3-bromobutyric acid ($[M]_{546}^{24} +116.2^\circ$ (0.168 M, 2M HClO₄), lit.¹³ $[M]_{546}^{25} +116.5^\circ$ (0.163 M, 2M HClO₄)), in a yield of 71%; $[\alpha]_D^{24} +26.8^\circ$ (C 5.0, CHCl₃).^{6,14}

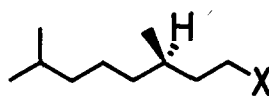
Synthesis of the key intermediate 2 was easily achieved by the S_N2 type ring-opening reaction of the lactone 10 with the Grignard reagent 11, prepared from 9 in 95% yield, in the presence of a copper(I) salt. Thus, when 11 was added to a mixture of 10 and copper(I) iodide (2 mol%) in THF-Me₂S (20:1) at -20 °C and the reaction mixture was allowed to warm to room temperature for 2.5 h, 2 was



4



5



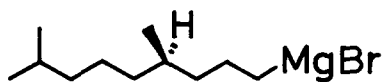
6 X = OH

8 X = CO₂H

7 X = Br

9 X = CH₂Br

10

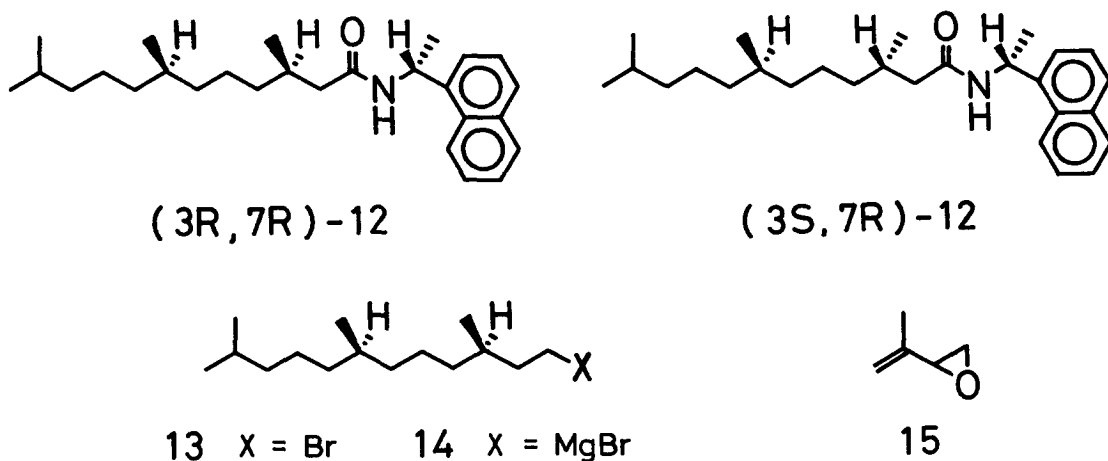


11

obtained in 90% yield; bp 180 °C (bath temp.)/1 mmHg; $[\alpha]_D^{24} +5.46^\circ$ (C 5.0, CHCl_3), lit.¹⁵ $[\alpha]_D +5.43^\circ$ (C 5.0, CHCl_3). The specific rotation indicated high optical purity of 2, and the enantiomeric purity of the asymmetric C_3 carbon was determined by TLC analysis of the diastereomeric amide (12), prepared from the acyl chloride of 2 and (R)-(+)- α -(1-naphthyl)ethylamine. Separation of 12 by silica gel TLC (C_6H_6 :AcOEt = 8:1) gave two components in a ratio of 96:4, corresponding to the (3R,7R)- and (3S,7R)-amides, respectively. Thus, the configuration of the β -carbon of (S)-3-bromobutyric acid was transformed into 2 with net retention of 96% through the two step inversion (cyclization to 10 and the ring-opening reaction of 10). Since the transformation of 4 to 2 seemed to proceed without racemization, the enantiomeric purity of the asymmetric C_3 and C_7 carbon of 2 could be determined to be 96% and 100% R, respectively.

Stereoselective introduction of a prenyl alcohol moiety with an E-configuration was readily achieved by a copper-catalyzed $\text{S}_{\text{N}}2'$ type reaction of Grignard reagent with the oxirane 15.^{7,16} The acid 2 upon reduction with lithium aluminum hydride gave the alcohol 3, followed by the treatment with hydrogen bromide to afford the corresponding bromide 13 in 87% yield; bp 132 °C/0.5 mmHg; $[\alpha]_D^{24} -4.44^\circ$ (C 5.0, CHCl_3). The Grignard reagent (14), prepared from 13 and magnesium by refluxing in THF in 76% yield, was added to a solution of 15 and copper(I) iodide (5 mol%) in THF- Me_2S (20:1) at -20 °C. After the reaction mixture was allowed to warm to 0 °C for 2 h, phytol (1) was obtained in 95% yield in a ratio of E to Z as 97:3.¹⁷

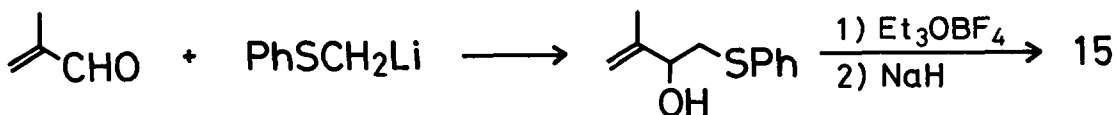
As mentioned above, the synthesis of (R,R)-phytol with highly stereochemical purity with regard to both absolute and geometrical configurations could be achieved by utilizing both copper(I) catalyzed Grignard reactions; the $\text{S}_{\text{N}}2$ type ring-opening reaction of (R)- β -methyl- β -propiolactone and the stereoselective $\text{S}_{\text{N}}2'$ type one of isopropenyl oxirane.



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References and Notes

1. J. W. Scott, F. T. Bizzaro, D. R. Parrish, and G. Saucy, *Helv. Chim. Acta*, **59**, 290 (1976); H. Mayer, P. Schudell, R. Rüegg, and O. Isler, *ibid.*, **67**, 650 (1963), and references cited therein.
2. Y. Naruta, *J. Org. Chem.*, **45**, 4097 (1980), and references cited therein.
3. J. W. K. Burrell, R. F. Garwood, L. M. Jackman, E. Oskay, and B. C. L. Weedon *J. Chem. Soc. (C)*, **1966**, 2144.
4. K. Sato, S. Mizuno, and M. Hirayama, *J. Org. Chem.*, **32**, 177 (1967); K. Sato and Y. Kurihara, *Yuki Gosei Kagaku Kyokai Shi*, **20**, 824 (1962), and references cited therein.
5. P. Karrer, A. Geiger, H. Rentschler, E. Zbinden, and A. Kugler, *Helv. Chim. Acta*, **26**, 1741 (1943).
6. T. Sato, T. Kawara, A. Nishizawa, and T. Fujisawa, *Tetrahedron Lett.*, **21**, 3377 (1980).
7. Stereoselective reaction of diorganocuprate with **15** was reported; R. J. Anderson, *J. Am. Chem. Soc.*, **92**, 4978 (1970).
8. C. Fuganti and P. Graselli, *J. Chem. Soc., Chem. Commun.*, **1979**, 995; N. Cohen, W. F. Eichel, R. J. Lopresti, C. Neukom, and G. Saucy, *J. Org. Chem.*, **41**, 3505 (1976); R. Zell, *Helv. Chim. Acta*, **62**, 474 (1979).
9. K. Chan, N. Cohen, J. P. De Noble, A. C. Specian, Jr., and G. Saucy, *J. Org. Chem.*, **41**, 3497 (1976).
10. B. M. Trost and T. P. Klun, *J. Am. Chem. Soc.*, **103**, 1864 (1981); J. Takahashi, K. Mori, and M. Matsui, *Agric. Biol. Chem.*, **43**, 1605 (1979).
11. C. G. Overberger and J. K. Weise, *J. Am. Chem. Soc.*, **90**, 3525 (1968).
12. J. R. Ruhoff, *Org. Synth.*, Coll. Vol. II. 292 (1950).
13. A. R. Olson and R. J. Miller, *J. Am. Chem. Soc.*, **60**, 2687 (1938).
14. J. R. Shelton, D. E. Agostini, and J. B. Lando, *J. Polym. Sci., A-1*, **9**, 2789 (1971).
15. D. Valentine, Jr., K. K. Chan, C. G. Scott, K. K. Johnson, K. Toth, and G. Saucy, *J. Org. Chem.*, **41**, 62 (1976).
16. Although the oxirane **15** is a useful reagent, there are a few reports on the preparative method.^{7,18} It was prepared by the modified procedure of Shanklin et al.¹⁹ 3-Methyl-1-phenylthio-3-buten-2-ol, prepared from methacrylaldehyde and phenylthiomethyl lithium, was treated with Et₃OBF₄ in CH₂Cl₂ at room



- temperature for 12 h. The solvent was exchanged by diglyme, and then a solution of the sulfonium salt was added to a suspension of NaH in diglyme at 0 °C under vacuum (0.5 v1 mmHg). The produced oxirane was collected in a trap cooled by liquid N₂, and distillation afforded pure **15** in 29% yield; bp 84 °C.
17. The ratio of E:Z was determined by capillary glpc analysis (F.F.A.P. 50 m) of phytol acetate, prepared from **1** and acetic anhydride in pyridine.
 18. P. M. Savu and J. A. Katzenellenbogen, *J. Org. Chem.*, **46**, 239 (1981).
 19. J. R. Shanklin, C. R. Johnson, J. Allinger, and R. M. Coates, *J. Am. Chem. Soc.*, **95**, 3429 (1973).

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